

Cryogenics Hands-on Training

Part 2: Liquid Oxygen/Eddy Current/Electrical Properties

The 7th Asian School on Superconductivity and Cryogenics for Accelerator

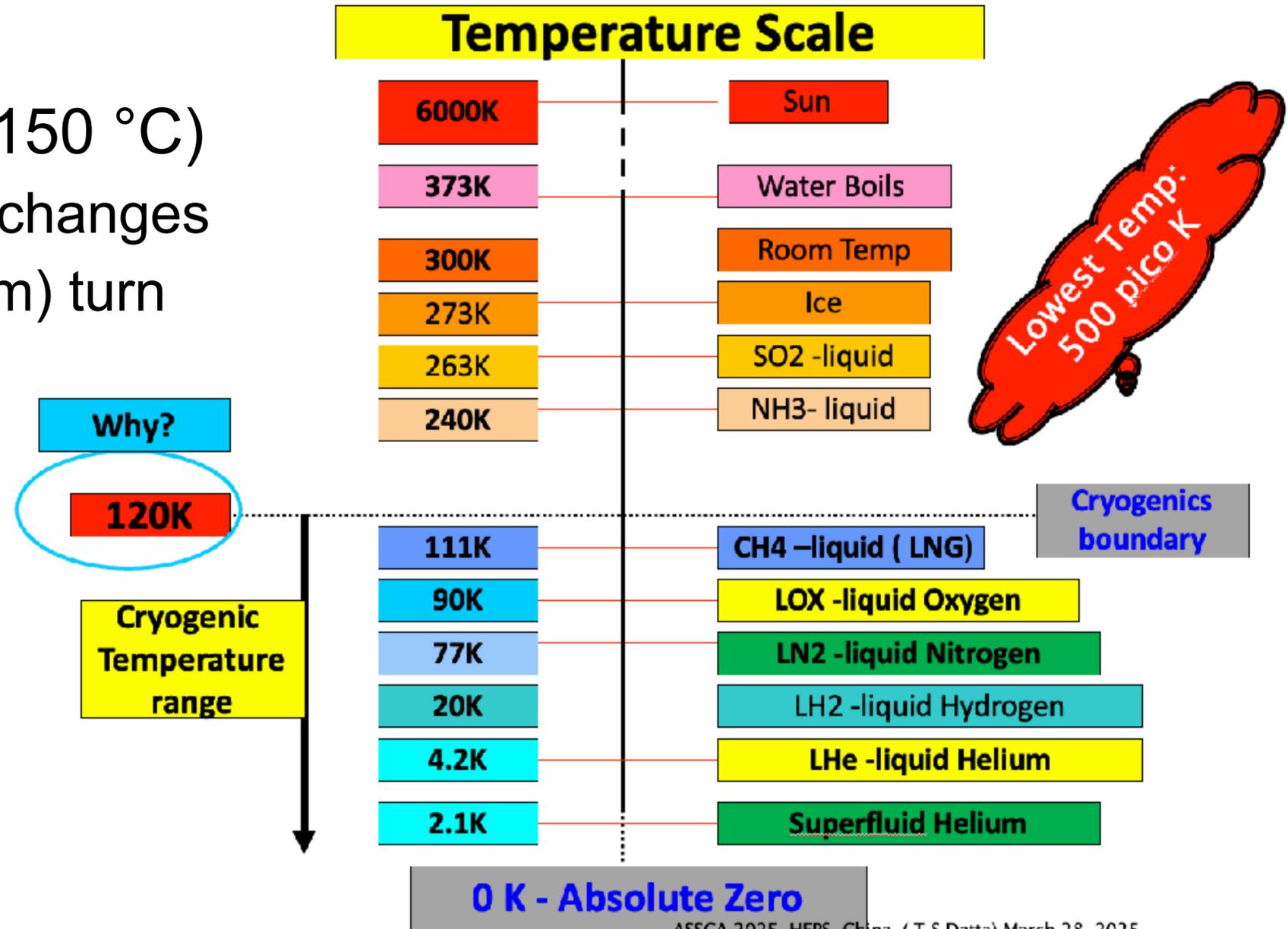
Feb 1 – 9, 2026

Soohyung Lee (Korea University)



What is Cryogenics?

- Cryogenics is a branch of physics that studies what happens to things at very low temperature.
- Cryogenic temperature: $< 123 \text{ K}$ (or $< -150 \text{ }^\circ\text{C}$)
 - “Normal” behavior of materials drastically changes
 - Gases (oxygen, nitrogen, hydrogen, helium) turn into liquids
- Many lectures in ASSCA2026
 - Cryogenics I (Prof. Sangkwon Jeong)
 - Cryogenics II (Prof. Rui Ge)
 - Cryogenics III (Prof. Hirotaka Nakai)
 - Cryogenics IV (Prof. Xilong Wang)
 - Cryogenics V (Dr. Seo Jung Kim)



ASSCA 2025, HEPS - China (T S Datta) March 28, 2025

Prof. T. S. Datta (ASSCA2025)

SAFETY WARNINGS

- We are using LN₂ today
 - Very common (78% of air, cheap) → good for simple coolings
- It's **VERY COLD** (77 K = -196 °C). **Be aware all the time, wear protection gears** (gloves and glasses)



Today's Hands-on Trainings

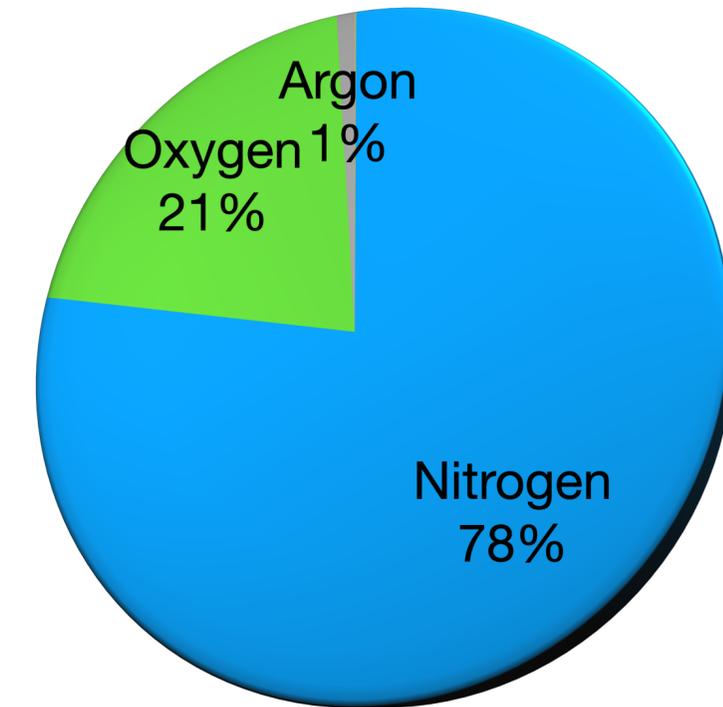
- #1: Liquid oxygen
 - Producing some liquid oxygen and confirming it
- #2: Eddy current
 - Feeling the effects of eddy current with various materials
- #3: Temperature coefficients of resistance and capacitance
 - Measuring resistance and capacitance and obtaining temperature coefficients of resistance and capacitance
- We'll do those in the accelerator lab (1st floor)

#1: Liquid Oxygen

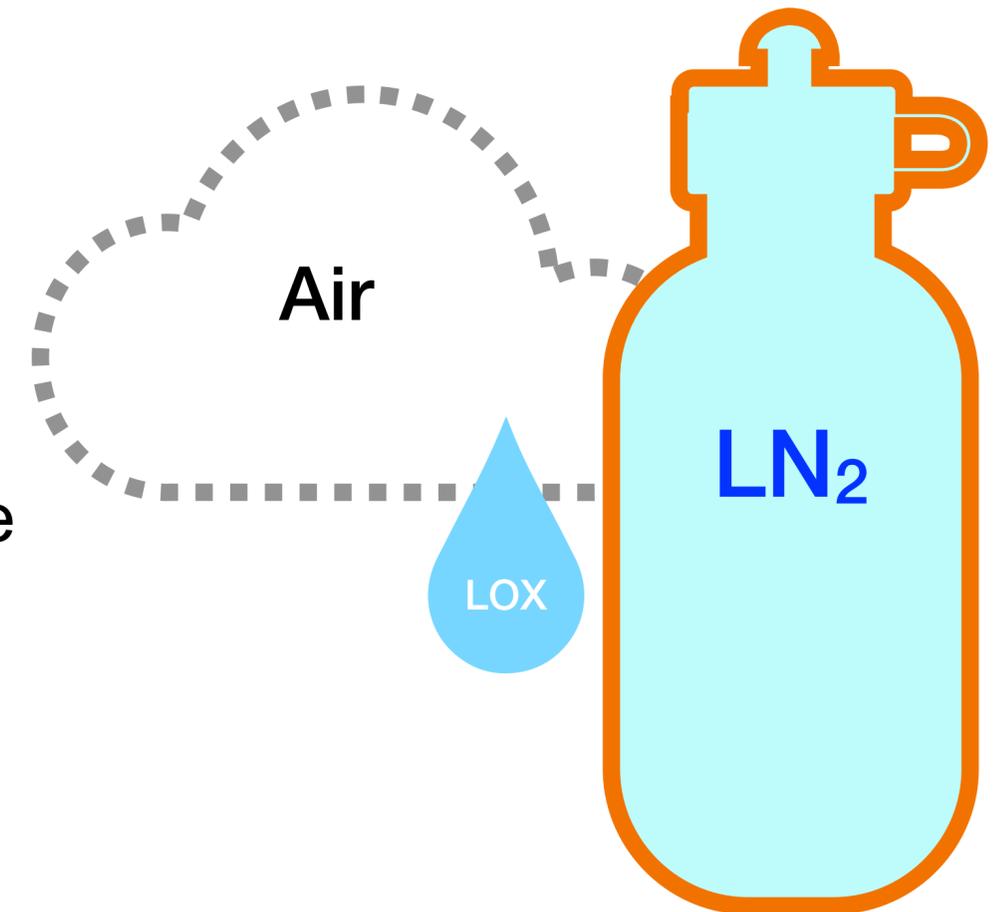
Liquid Oxygen

- Oxygen is the 2nd most common element in the air

Element	Portion [%]	Boiling Point [K]
Nitrogen (N ₂)	78.084	77.4
Oxygen (O ₂)	20.946	90.2
Argon (Ar)	0.9340	87.3
Carbon dioxide (CO ₂)	0.04	194.7



- Liquid nitrogen (LN₂) will provide a lower temperature than the boiling point of oxygen
 - Oxygen in the air will condense to a liquid (LOX) at the temperature
 - LOX can be confirmed by its properties:
 - Pale blue
 - Paramagnetic: weakly attracted by a magnet
 - Oxidant: Don't do this...



Liquid Oxygen

- What we are going to do:
 - Pouring an amount of LN₂ into a copper cone
 - Wait for a while until the copper cone is cooled down enough
 - At some point, you'll see some liquid drops from the bottom
 - Place a magnet if the liquid is affected by the magnetic



#2: Eddy Current

Eddy Current

- Varying magnetic field induces a current inside a conductor → “Eddy current”
 - Eddy current generates a secondary magnetic field against the varying magnetic field (Lenz’s law)

- Many applications in our real-life



Eddy Current

- Maxwell's equation (in vacuum)

$$\nabla \cdot \vec{E} = \frac{\rho}{\epsilon_0}$$

$$\nabla \times \vec{E} = -\frac{\partial \vec{B}}{\partial t}$$

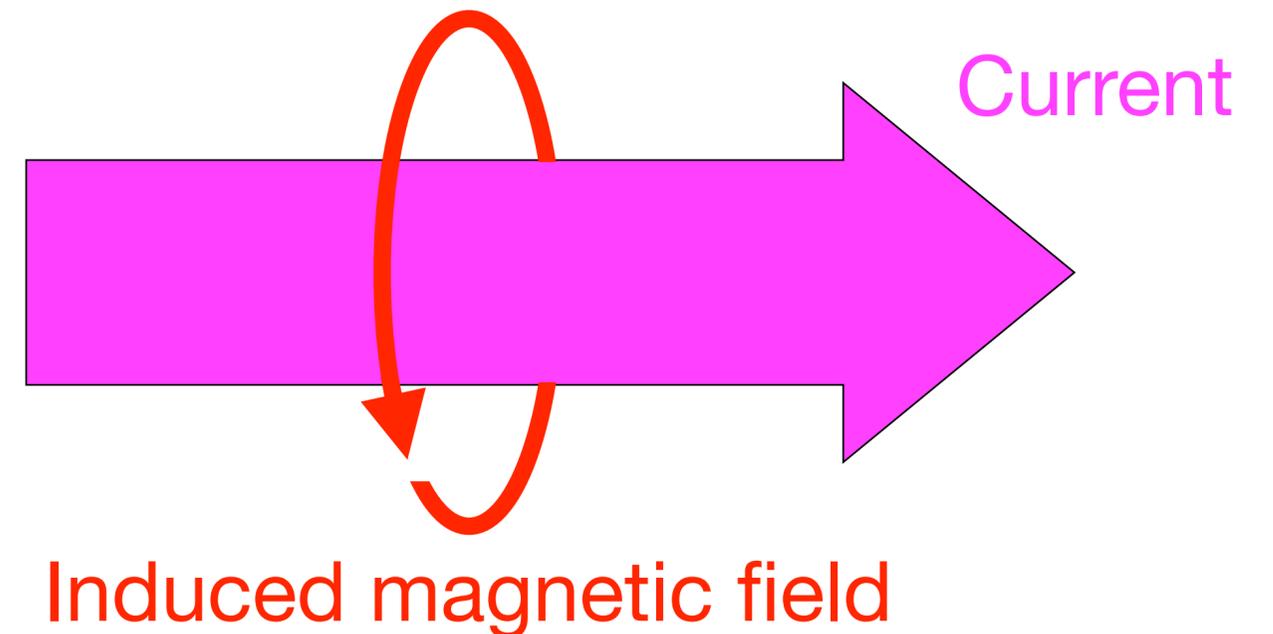
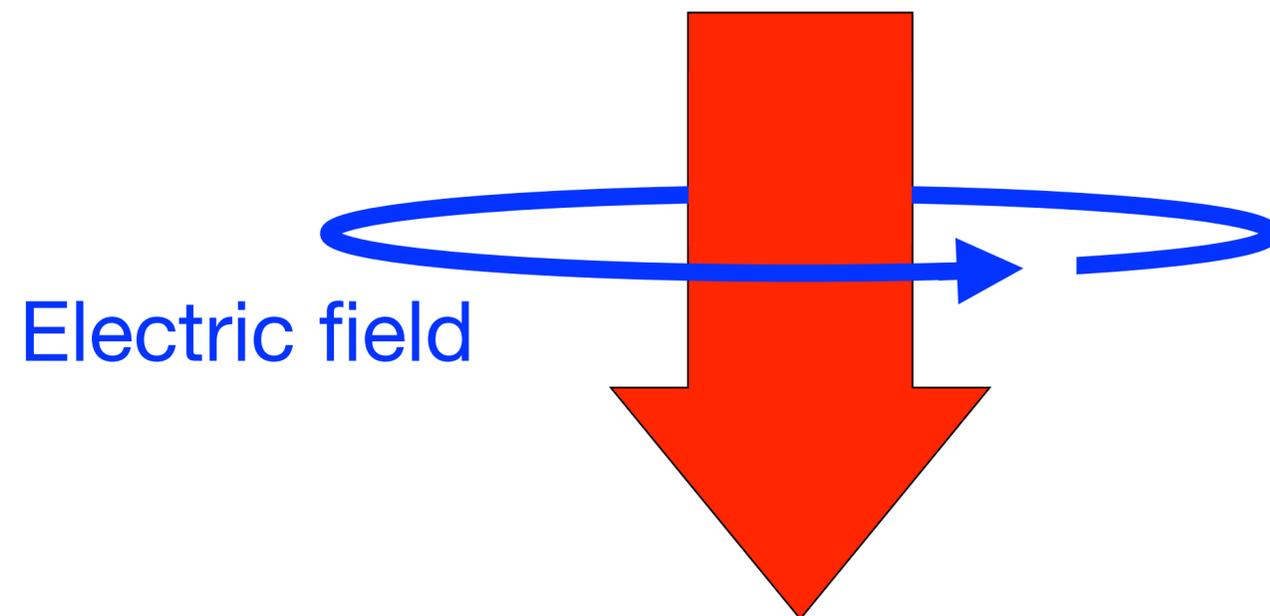
Faraday's law

$$\nabla \cdot \vec{B} = 0$$

$$\nabla \times \vec{B} = \mu_0 \vec{J} + \epsilon_0 \mu_0 \frac{\partial \vec{E}}{\partial t}$$

Ampere's law

Varying (increasing) magnetic field



Eddy Current

- If the varying magnetic field is surrounded by a conductor,

$$\nabla \times \vec{E} = -\frac{\partial \vec{B}}{\partial t}$$

Faraday's law

$$\nabla \times \vec{B} = \mu_0 \vec{J} + \epsilon_0 \mu_0 \frac{\partial \vec{E}}{\partial t}$$

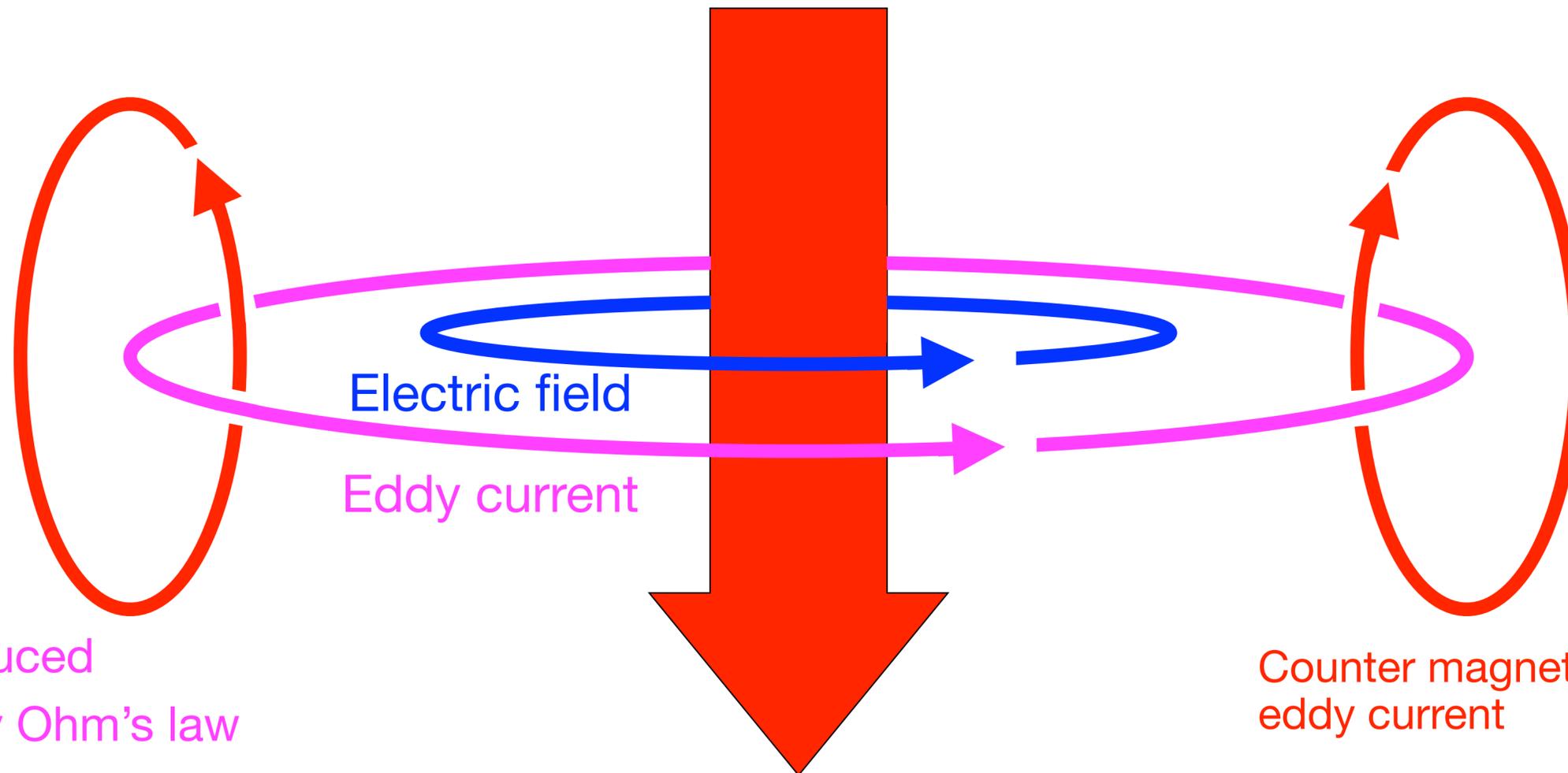
Ampere's law

Varying (increasing) magnetic field

$$\vec{J} = \sigma \vec{E}$$

Ohm's law

Eddy current is induced in the conductor by Ohm's law



Counter magnetic field is induced by the eddy current

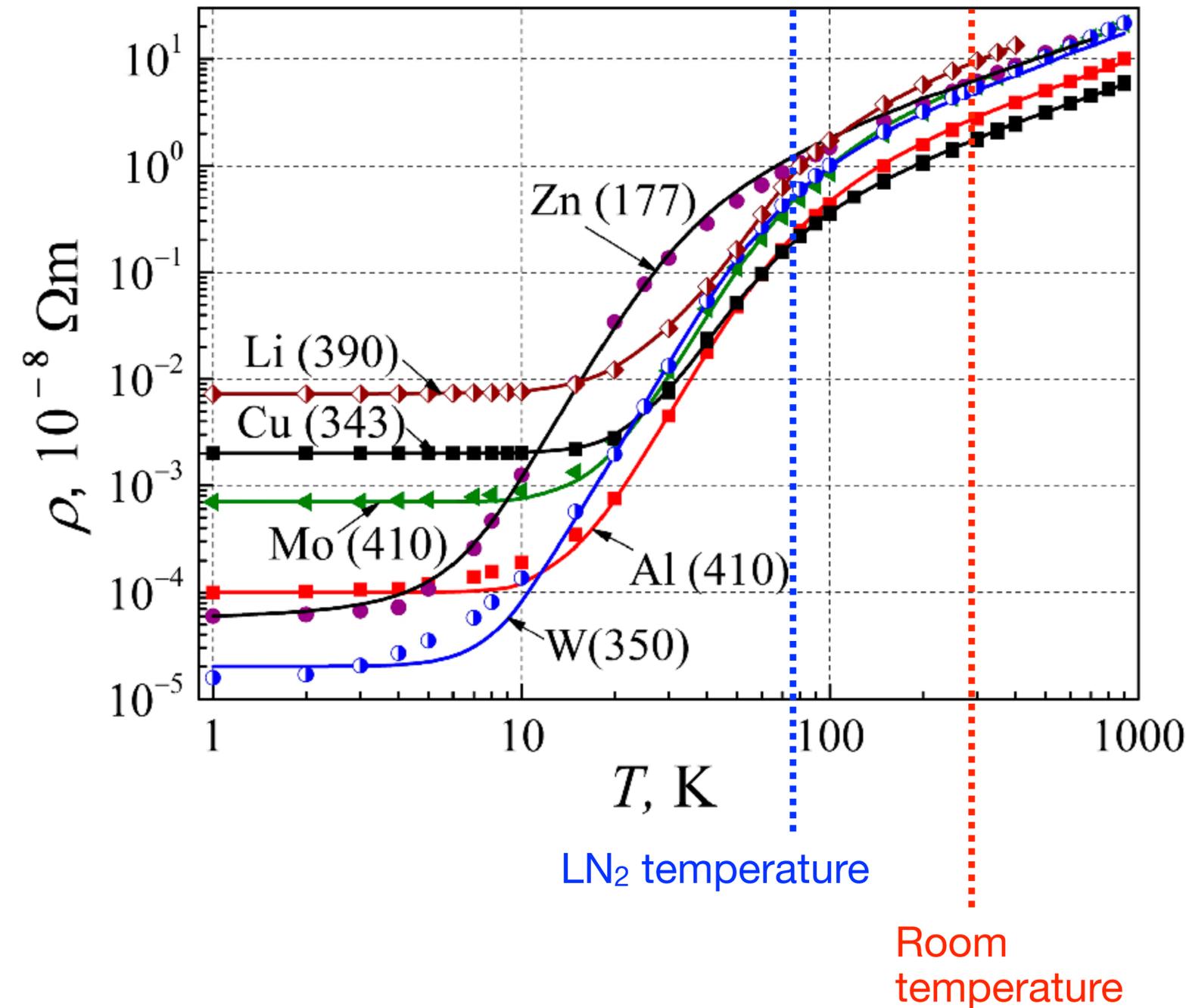
Eddy Current

- Ohm's law tells us that a material with a higher conductivity (=less resistance) yields a higher eddy current with a given electric field

$$\vec{J} = \sigma \vec{E}$$

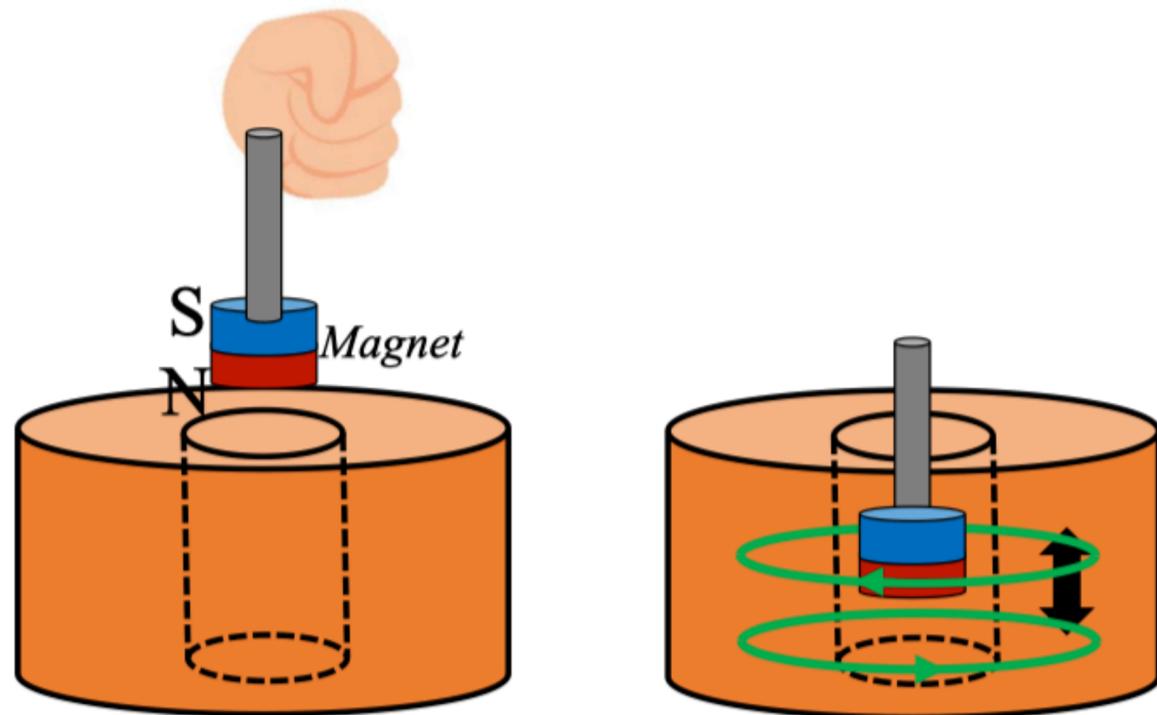
- Resistivity of metal usually decreases as the temperature decreases
 - e.g. Copper: $\rho_{300\text{ K}} / \rho_{77\text{ K}} \approx 8$
- Thus, stronger counter magnetic field at lower temperatures

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Eddy Current

- What we are going to do:
 - Pushing and pulling a magnet inside conductors at a room-temperature
 - Copper and aluminum
 - Do the same with cryogenic copper
 - Feel the resistive force



#3: Temperature Coefficients of Resistance and Capacitance

Temperature Coefficients of Resistance and Capacitance

- As seen, resistivity of a material changes over its temperature
 - Extreme example: superconductors ($\rho_{\text{Nb}} = 0$ at $T_c < 9.25$ K)
- Temperature coefficient of resistance (TCR)

$$R = R_0 [1 + \alpha(T - T_0)]$$

R : resistance at a temperature T
 R_0 : resistance at a reference temperature T_0
 α : temperature coefficient of resistance

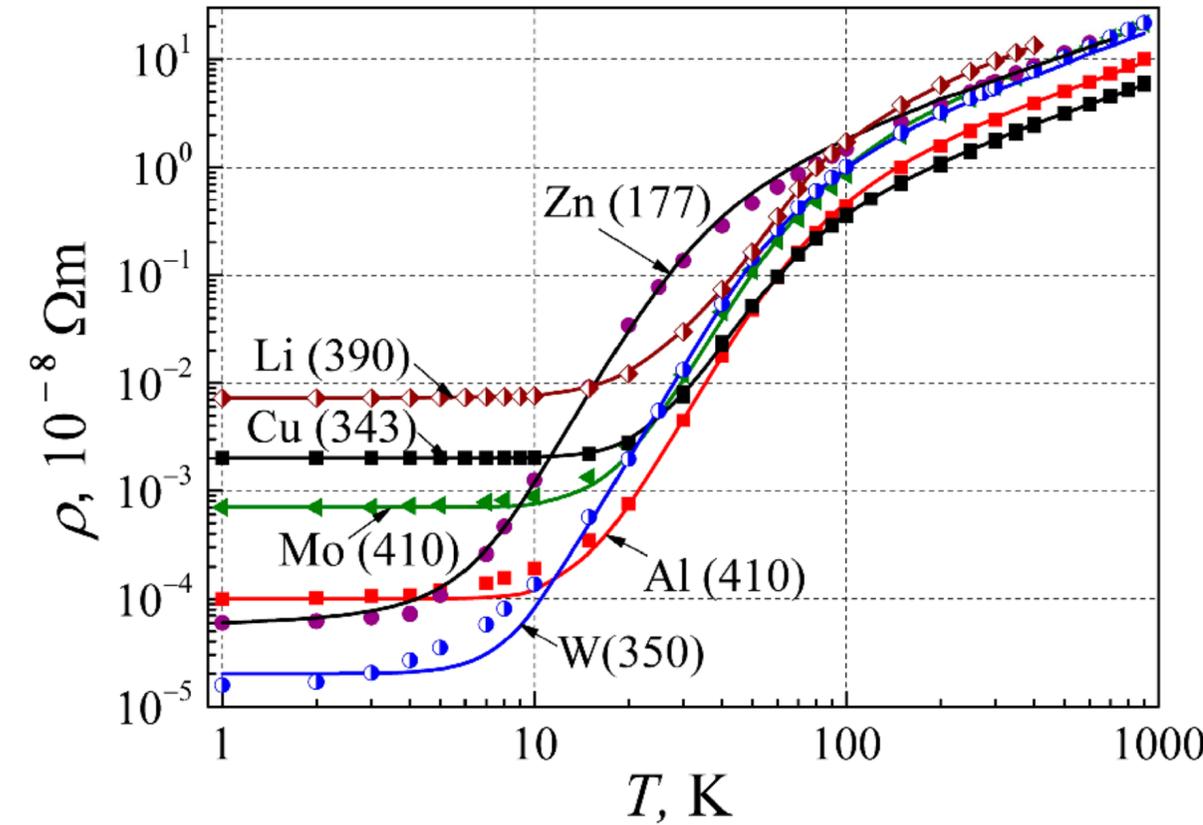
- Capacitance also changes over temperature
 - Changing of dielectric material behavior in capacitors
 - e.g. electrolytic capacitors: electrolytes are frozen → drastically decreasing capacitance

- Temperature coefficient of capacitance (TCC)

$$C = C_0 [1 + \alpha(T - T_0)]$$

C : resistance at a temperature T
 C_0 : resistance at a reference temperature T_0
 α : temperature coefficient of capacitance

V. Palenskis, *Metals* **2024**, 526 (2024)



By measuring resistances (or capacitances), you can extract TCR (or RCC)

Temperature Coefficients of Resistance and Capacitance

- What we are going to do: Obtaining TCR and TCC of various resistors/capacitors

- Resistors

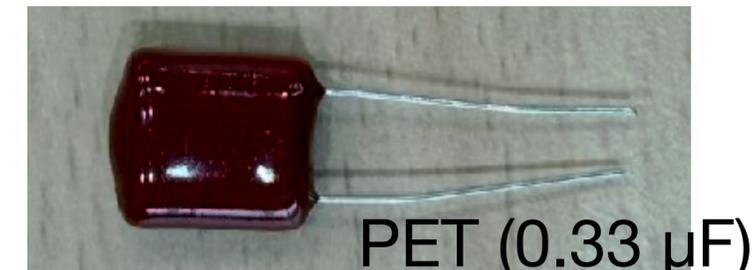
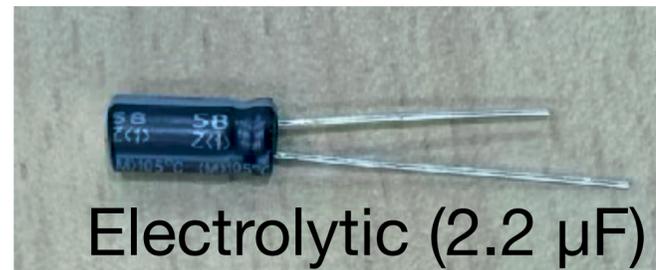
- Metal film resistors



- Measure the resistance of them at a room-temperature and a cryogenic temperature
- Estimate TCR and interpret the measurement

- Capacitors

- Electrolytic capacitor (aluminum)
- Metallized polyester capacitor (PET)



- Measure the capacitance of them at a room-temperature and a cryogenic temperature
- Estimate TCC and interpret the measurement

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